

Fig. 1. One asymmetrical unit of β_1 - K_2UF_6 . One-fourth unit cell is outlined.

Table 3. Observed and calculated structure factors for β_1 - K_2UF_6 , F_o , F_c and $\alpha \times 10$

L	F _o	F _c	ALPHA	L	F _o	F _c	ALPHA	L	F _o	F _c	ALPHA	L	F _o	F _c	ALPHA	L	F _o	F _c	ALPHA	L	F _o	F _c	ALPHA						
0	0	0	L	1	456	443	95	2	246	241	159	3	188	162	39	1	4	L	4	201	207	220	3	3	L				
1	247	241	179	2	235	239	169	3	215	213	167	4	227	220	134	0	368	343	47	0	252	233	126	1	364	327	67		
2	469	580	70	3	295	265	131	0	5	L	0	1	2	L	1	135	136	160	0	252	233	126	1	226	227	240			
3	170	192	199	4	180	168	203	0	258	249	157	0	270	278	303	2	296	289	72	1	324	318	151	2	292	279	67		
4	295	292	117	0	3	L	0	1	262	254	151	1	439	408	28	0	197	210	212	2	202	198	150	3	178	173	249		
0	1	L	0	556	521	80	2	225	215	172	2	210	237	275	0	197	210	212	2	4	L	0	204	199	168	0	176	179	176
0	410	385	113	1	293	279	148	0	6	L	3	296	271	86	1	260	253	136	0	204	199	168	0	176	179	176			
1	369	400	106	2	412	400	98	0	292	303	123	4	169	166	257	2	2	L	1	275	276	168	2	162	162	190			
2	263	296	137	3	212	202	183	1	194	180	206	0	1	3	L	0	406	382	240	2	178	179	163	3	5	L	0		
3	259	267	142	4	266	239	138	0	1	L	0	283	267	181	1	198	209	271	3	214	210	183	0	154	173	218			
4	196	188	183	0	4	L	0	528	451	48	1	323	322	96	2	287	315	227	2	5	L	0	4	4	L	0			
0	2	L	0	303	285	142	1	293	201	186	2	224	229	190	3	159	169	261	2	209	229	210	0	241	242	107			
0	313	277	155	1	281	276	145	-2	343	349	80	3	240	236	133														

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The crystal structure of γ - Na_2ZrF_6 .* By GEORGE BRUNTON, *Reactor Chemistry Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37830, U.S.A.*

(Received 25 January 1969)

The crystal structure of γ - Na_2ZrF_6 is a new structure type with space group $P2_1/c$. The Zr ions are coordinated by $7F^-$ ions at distances of 2.012 to 2.167 Å. The structure is metastable below 460°C.

The complex fluoride Na_2ZrF_6 has four polymorphs (Barton, Grimes, Insley, Moore & Thoma, 1958) which can be distinguished from one another by their optical and X-ray properties. γ - Na_2ZrF_6 is biaxial positive; $2V=75^\circ$, $N_x=1.408$ and $N_z=1.412$. The equilibrium β - γ transition is at

505C and the γ - δ transition is at 460C. The crystal structure of γ - Na_2ZrF_6 is a new structure type.

Experimental

Single crystals of γ - Na_2ZrF_6 were obtained from a rapidly cooled ingot of the composition $NaF-ZrF_4-UF_4$ (55-40-5 mole%). The crystals were ground in a small air race, and

* Research sponsored by the U.S. Atomic Energy Commission under contract with Union Carbide Corporation.

Table 1. *The lattice and atomic parameters of γ -Na₂ZrF₆*

$a_0 = 5.5562$ (7), $b_0 = 5.4069$ (6), $c_0 = 16.073$ (2) Å; $\beta = 95.886$ (6)°. $\rho_c = 3.4728$ g.cm⁻³, $Z = 4$.

	$x \times 10^3$	$y \times 10^3$	$z \times 10^3$	$\beta_{11}^* \times 10^3$	$\beta_{22} \times 10^3$	$\beta_{33} \times 10^4$	$\beta_{12} \times 10^3$	$\beta_{13} \times 10^4$	$\beta_{23} \times 10^4$
Zr	176.1 (2)	448.0 (3)	98.90 (5)	21.0 (6)	18 (2)	20.0 (7)	0.8 (3)	4 (2)	1.0 (7)
Na(1)	326.7 (8)	499 (2)	407.7 (3)	26 (3)	25 (3)	27 (3)	-2 (2)	11 (6)	-4 (5)
Na(2)	246.7 (9)	962 (2)	247.7 (4)	26 (2)	33 (4)	34 (3)	4 (2)	24 (6)	15 (6)
F(1)	20 (2)	709 (2)	335.8 (5)	31 (3)	15 (4)	33 (3)	2 (3)	-4 (7)	2 (8)
F(2)	55 (2)	176 (2)	350.3 (5)	27 (3)	35 (4)	33 (3)	-7 (3)	14 (6)	10 (9)
F(3)	346 (2)	224 (2)	24.2 (5)	31 (3)	22 (4)	39 (3)	1 (3)	5 (7)	1 (9)
F(4)	598 (2)	229 (2)	364.8 (4)	30 (3)	28 (4)	27 (3)	3 (2)	11 (6)	7 (7)
F(5)	430 (2)	286 (2)	183.5 (4)	29 (3)	37 (5)	30 (3)	-1 (3)	-11 (7)	31 (9)
F(6)	871 (2)	328 (2)	11.6 (4)	23 (2)	17 (4)	32 (3)	-4 (2)	-9 (6)	9 (8)

* Coefficients in the temperature factor: $\exp [-(\beta_{11}h^2 + \beta_{22}k^2 + \beta_{33}l^2 + 2\beta_{12}hk + 2\beta_{13}hl + 2\beta_{23}kl)]$.
The number in parentheses is the standard error in terms of the last significant figure.

an ellipsoid $0.156 \times 0.156 \times 0.208$ mm was mounted on a computer operated Picker four-circle goniostat. The intensities of 489 independent reflections were measured with a scintillation-counter detector using unfiltered Cu $K\alpha$ radiation ($K\alpha_1 = 1.54050$ and $K\alpha_2 = 1.54434$) out to $145^\circ 2\theta$ by the 2θ scan technique. The linear absorption for Cu $K\alpha$ radiation by γ -Na₂ZrF₆ is 225.53 cm⁻¹ and an absorption correction was calculated for each reflection.

Refinement and results

The observed conditions for diffraction: hkl —no conditions, $h0l$ — $l = 2n$ and $0k0$ — $k = 2n$ are consistent for the space group $P2_1/c$ (14). The lattice parameters were refined from 17 Cu $K\alpha_1$ reflections in the 2θ range 92 – 110° , Table 1.

The structure was refined by iterative least-squares using a modification of the Busing, Martin & Levy (1962) computer program. The starting parameters were determined from a three-dimensional Patterson function, and the results are listed in Table 1. An extinction correction was made on F_c by the method suggested by Zachariasen (1967). The scattering factors for the ions were taken from Cromer & Waber (1965) and the values of $\Delta f' = -0.6$ and $\Delta f'' = 2.5$ electrons were used for the anomalous dispersion of Cu $K\alpha$ radiation by zirconium (Dauben & Templeton, 1955).

The quantity minimized by the least-squares program was $\sum w||sF_o| - |F_c||^2$ with weights, w , equal to the reciprocals of the variances which were estimated from the empirical equation:

$$\sigma^2(F_o^2) = [T + B + \{0.15(T - B)\}^2] / [A(Lp)^2]$$

where T = total counts, B = background counts, A = absorp-

tion correction and Lp = Lorentz-polarization (Brown & Levy, 1964); $\sigma(F_o) = \sigma(F_o^2)/2F_o$. Anisotropic temperature factors were calculated for all atoms (Table 1) and a stereoscopic pair of drawings of the structure of γ -Na₂ZrF₆ is shown in Fig. 1.

The discrepancy factor $R = \sum ||F_o| - |F_c|| / \sum |F_o| = 0.0536$ for 489 independent reflections. The standard deviation of an observation of unit weight, $[\sum w(F_o - F_c)^2 / (n_o - n_v)]^{1/2}$ where n_o is the number of reflections and n_v the number of variables, is 1.307. The interatomic distances are listed in Table 2, and the observed and calculated structure factors are listed in Table 3.

Table 2. *Interatomic distances γ -Na₂ZrF₆*

Zr—F(4)	2.012 (7) Å	Na(1)—F(3)	2.218 (9) Å
Zr—F(3)	2.019 (8)	Na(1)—F(1)	2.249 (9)
Zr—F(2)	2.024 (7)	Na(1)—F(4)	2.264 (9)
Zr—F(5)	2.035 (7)	Na(1)—F(3)	2.35 (1)
Zr—F(1)	2.058 (8)	Na(1)—F(2)	2.41 (1)
Zr—F(6)	2.138 (7)	Na(1)—F(6)	2.53 (2)
Zr—F(6)	2.167 (6)	Na(1)—F(5)	2.62 (2)
Na(2)—F(5)	2.211 (8)	Na(2)—F(1)	2.429 (9)
Na(2)—F(1)	2.302 (8)	Na(2)—F(4)	2.447 (9)
Na(2)—F(5)	2.33 (1)	Na(2)—F(2)	2.66 (1)
Na(2)—F(2)	2.369 (9)	Na(2)—F(4)	2.922 (8)
F(1)—F(5)	2.515 (9)	F(1)—F(6)	2.54 (1)
F(1)—F(2)	2.54 (2)	F(1)—F(2)	2.89 (2)
F(1)—F(2)	2.98 (2)	F(2)—F(5)	2.595 (9)
2[F(2)—F(6)]	2.90 (1)	F(3)—F(5)	2.569 (9)
F(3)—F(6)	2.685 (9)	F(3)—F(6)	2.74 (2)
F(3)—F(6)	3.000 (9)	F(4)—F(5)	2.52 (2)
F(4)—F(6)	2.665 (9)	F(4)—F(5)	2.968 (8)

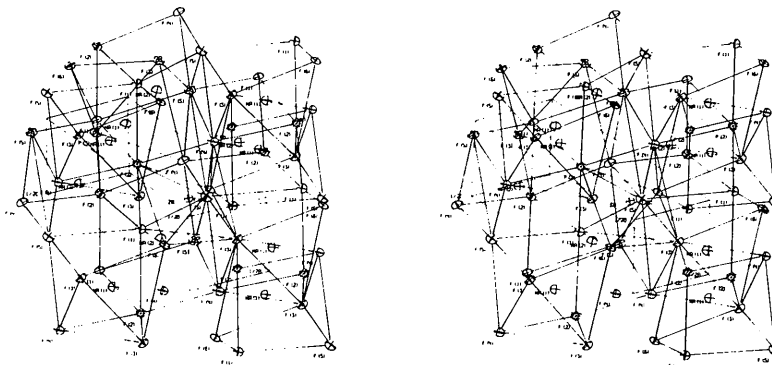


Fig. 1. The crystal structure of γ -Na₂ZrF₆. The four nearest neighbor Na(1) and Na(2) polyhedra are shown around one Zr polyhedron. One fourth unit-cell outlined.

